

Reactions of Benzimidazole-2-thiol with *N*-Alkyl-1-aza-1,3-enynes

M. V. Karpov, A. V. Garabadzhiu, P. B. Davidovich, and A. N. Belyaev

St. Petersburg State Institute of Technology, Moskovskii pr. 26, St. Petersburg, 198013 Russia
e-mail: mihkarpov@gmail.com

Received April 7, 2014

Abstract—Reactions of *N*-alkyl-1-aza-1,3-enynes with benzimidazole-2-thiol in anhydrous dimethylformamide have been studied. The structure of the obtained compounds has been elucidated from the NMR data.

Keywords: benzimidazole-2-thiol, *N*-alkyl-1-aza-1,3-enyne, heterocyclization

DOI: 10.1134/S1070363214080209

1-Aza-1,3-enynes are convenient and readily available synthons for synthesis of organic sulfur-containing polynuclear heterocyclic compounds [1–6]. Recently, considerable attention has been focused on the heterocyclization reactions of 1-aza-1,3-enynes resulting from addition of various nucleophilic reagents [3]. Benzimidazole-2-thiol derivatives readily participate in the substitution and addition (at activated double or triple bond) reactions resulting in formation of heterocyclic or acyclic products [7, 8]. Benzimidazole-2-thiol and its derivatives are known for anxiolytic activity [9]; some of them inhibit chymases (enzymes responsible for activation of interleukin-1 β , serving among the major mediators of non-specific inflammatory forms of protection) [10].

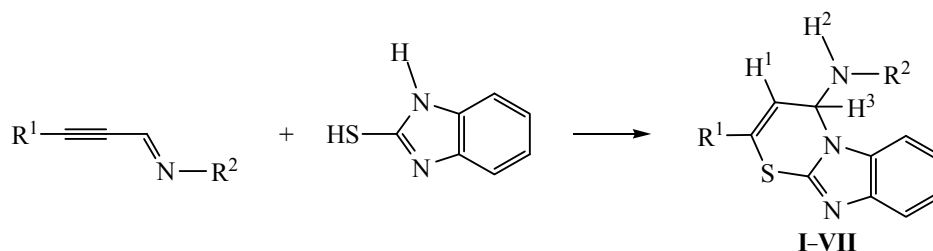
It was previously shown that interaction of *N*-*tert*-butyl-1-aza-1,3-enynes with benzimidazole-2-thiol led to formation of polynuclear heterocyclic sulfur compounds, benzo[4,5]imidazole[2,1-*b*][1,3]thiazine-4-ols

[11]. The reaction occurred in methanol and was accompanied by heterolytic cleavage of *tert*-butylamine; in the case of silicon-containing 1-aza-1,3-enyne, desilylation was observed. However, addition of adenine-8-thiol to the silicon-containing 1-aza-1,3-enynes proceeded in absolute DMF without desilylation and allowed to preserve the secondary amino group [12].

In this work we studied reactions of benzimidazole-2-thiol with *N*-alkyl-1-aza-1,3-enynes (obtained via condensation of phenylpropionic aldehyde and 4,4-dimethylpent-3-ynal) with aliphatic amines (*tert*-butylamine, cyclohexylamine, and benzylamine) [1].

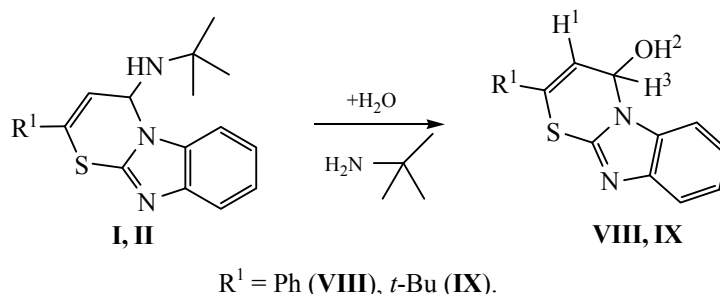
Similarly to the above-mentioned result [12], we found that using anhydrous DMF as solvent prevented hydrolysis and preserved the amine moiety. The obtained compounds **I–VII** were crystalline substances; their structure was confirmed by NMR spectroscopy (Scheme 1).

Scheme 1.



R¹ = Ph, R² = *t*-Bu (**I**); R¹ = *t*-Bu, R² = *t*-Bu (**II**); R¹ = Me₃Si, R² = *t*-Bu (**III**); R¹ = *t*-Bu, R² = cyclo-C₆H₁₁ (**IV**); R¹ = Me₃Si, R² = cyclo-C₆H₁₁ (**V**); R¹ = *t*-Bu, R² = CH₂Ph (**VI**); R¹ = Me₃Si, R² = CH₂Ph (**VII**).

Scheme 2.



In ^1H NMR spectrum of **II**, doublet at 6.15 ppm ($^3J_{\text{HH}}$ 5.82 Hz) was assigned to the H^1 atom. The signal of H^1 proton was observed at 2.34 ppm. The H^3 atom of 1,3-thiazine ring resonated as multiplet in the range of 6.01–6.05 ppm. The singlets at 1.11 and 1.30 ppm belonged to the protons of two *tert*-butyl groups. Signals of the aromatic protons were registered as three multiplets in their typical area.

^1H NMR spectra of trimethylsilyl derivatives **III**, **V**, and **VII** contained distinct singlets of nine protons in the range of 0.25–0.30 ppm [12].

According to [4, 13], addition of azoles at multiple bond of 1-aza-1,3-enynes proceeded much less readily than in the cases of aromatic thiols [1]. Apparently, thiol group was the first to react with 1-aza-1,3-enynes, and then the pyrrole nitrogen atom of benzimidazole-2-thiol followed.

Compounds **I** and **II** were hydrolyzed in aqueous ethanol (1 : 1) solution, the process being accompanied with substitution of *tert*-butylamino group to hydroxyl moiety.

^1H NMR spectrum of **IX** contained signals of the H^1 (6.00 ppm, J_{HH} 4.36 Hz), H^2 (6.45 ppm), and H^3 (6.78 ppm, J_{HH} 7.26 Hz) atoms. The singlet at 1.35 ppm was assigned to the *tert*-butyl protons. Aromatic protons resonated in their usual spectral region. No signals of the *tert*-butylamino group were observed in the spectrum, evidencing about its cleavage. Compounds **VIII** and **IX** were previously described in [11], but they were obtained directly by reacting *N*-*tert*-butyl-1-aza-1,3-enynes with benzimidazole-2-thiol, without isolation of the intermediates **I** and **II** (Scheme 2).

A notable feature of the described reaction is the solvent effect. In particular, performing the reaction in anhydrous DMF gives 2-substituted 4*H*-benzo[4,5]-imidazole[2,1-*b*][1,3]thiazine-4-amines **I–VII**, whereas with methanol as solvent it yields 2-sub-

stituted 4*H*-benzo[4,5]imidazole[2,1-*b*][1,3]thiazine-4-ols **VIII** and **IX**.

EXPERIMENTAL

NMR spectra ($\text{DMSO-}d_6$) were recorded with the Varian XL-300 spectrometer [300.13 MHz (^1H), 75.45 MHz (^{13}C)] relative to the signal of residual solvent protons. Analytical pure grade reagents were used for the syntheses.

***N*-*tert*-Butyl-2-phenyl-4*H*-[1,3]thiazino[3,2-*a*]benzimidazole-4-amine (I).** A mixture of a solution of 0.01 mol of phenylpropionic aldehyde *tert*-butylimine in 50 mL of anhydrous DMF and 0.01 mol of benzimidazole-2-thiol was stirred during 2 h. After the solvent removal, oily residue was crystallized from toluene–petroleum ether mixture. Pale-brown crystals were filtered off and dried in air. Yield ~75%. ^1H NMR spectrum, δ , ppm: 1.25 s (9H), 1.97 br.s (1H), 6.11 m (1H), 6.51 d (1H, $^3J_{\text{HH}}$ 5.8 Hz), 7.25–7.54 m (9H). ^{13}C NMR spectrum, δ_{C} , ppm: 30.48, 50.17, 62.71, 110.80, 117.55, 120.28, 121.00, 121.50, 121.80, 126.10, 127.20, 128.10, 128.70, 129.05, 129.31, 133.33, 143.25, 145.00. Found, %: C 71.6; H 6.5; N 12.5. $\text{C}_{20}\text{H}_{21}\text{N}_3\text{S}$. Calculated, %: C 71.6; H 6.3; N 12.5.

***N*,2-Di-*tert*-butyl-4*H*-[1,3]thiazino[3,2-*a*]benzimidazole-4-amine (II).** A mixture of a solution of 0.012 mol of *N*-(4,4-dimethylpent-2-yn-1-ylidene)-2-methylpropane-2-amine in 50 mL of anhydrous DMF and 0.010 mol of benzimidazole-2-thiol was incubated during 12 h. The precipitated crystals were filtered off and dried in air. Yield ~50%, mp 155°C. ^1H NMR spectrum, δ , ppm: 1.11 s (9H), 1.30 s (9H), 2.34 br.s (1H), 6.03 m (1H), 6.15 d (1H, $^3J_{\text{HH}}$ 5.82 Hz), 7.12–7.17 m (2H), 7.46–7.49 m (1H), 7.58–7.60 m (1H). ^{13}C NMR spectrum, δ_{C} , ppm: 29.03, 30.38, 36.40, 49.98, 62.20, 110.53, 117.11, 117.40, 120.77, 121.58, 133.28, 140.02, 143.14, 145.81. Found, %: C 68.4; H 7.9; N 13.7. $\text{C}_{18}\text{H}_{25}\text{N}_3\text{S}$. Calculated, %: C 68.5; H 8.0; N 13.3.

***N*-tert-Butyl-2-trimethylsilyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-amine (III)** was obtained similarly. Yield 78%, mp 166°C. ^1H NMR spectrum, δ , ppm: 0.29 s (9H), 1.17 s (9H), 1.85 d (1H, $^3J_{\text{HH}}$ 10.18 Hz), 5.94–5.99 m (1H), 6.46 d (1H, $^3J_{\text{HH}}$ 5.09 Hz), 7.24–7.27 m (2H), 7.48–7.50 m (1H), 7.67–7.69 m (1H). ^{13}C NMR spectrum, δ_{C} , ppm: 2.26, 30.40, 50.02, 62.85, 110.61, 117.35, 120.75, 121.70, 128.19, 129.62, 133.30, 142.53, 145.36. Found, %: C 61.4; H 7.5; N 11.7. $\text{C}_{17}\text{H}_{25}\text{N}_3\text{SSi}$. Calculated, %: C 61.5; H 7.6; N 11.6.

***N*-Cyclohexyl-2-tert-butyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-amine (IV)** was obtained similarly. Yield 65%, mp 128°C. ^1H NMR spectrum, δ , ppm: 0.76–1.65 m (10H), 1.31 s (9H), 2.40 br.s (1H), 2.61 br.s (1H), 5.95 d (1H, $^3J_{\text{HH}}$ 5.09 Hz), 6.10 m (1H), 7.10–7.18 m (2H), 7.47–7.50 m (1H), 7.61–7.64 m (1H). Found, %: C 70.3; H 7.9; N 12.4. $\text{C}_{20}\text{H}_{27}\text{N}_3\text{S}$. Calculated, %: C 70.3; H 8.0; N 12.3.

***N*-Cyclohexyl-2-trimethylsilyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-amine (V)** was obtained similarly. Yield 70%, mp 130°C. ^1H NMR spectrum, δ , ppm: 0.76–1.65 m (10H), 0.30 s (9H), 2.39 br.s (1H), 2.70 br.s (1H), 6.13 m (1H), 6.21 d (1H, $^3J_{\text{HH}}$ 4.36 Hz), 7.10–7.19 m (2H), 7.47–7.50 m (1H), 7.62–7.65 m (1H). ^{13}C NMR spectrum, δ_{C} , ppm: 1.62, 23.86, 24.59, 24.88, 25.32, 30.22, 30.73, 31.19, 32.57, 109.30, 120.35, 121.80, 168.00. Found, %: C 63.7; H 7.7; N 11.7. $\text{C}_{19}\text{H}_{27}\text{N}_3\text{SSi}$. Calculated, %: C 63.8; H 7.6; N 11.8.

***N*-Benzyl-2-tert-butyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-amine (VI)** was obtained similarly. Yield 70%, mp 148°C. ^1H NMR spectrum, δ , ppm: 1.25 s (9H), 2.39 br.s (1H), 3.30–3.33 m (1H), 3.50–3.53 m (1H), 3.57 br.s (1H), 5.90 d (1H, $^3J_{\text{HH}}$ 5.09 Hz), 6.22 m (1H), 7.10–7.21 m (7H), 7.49–7.51 m (1H), 7.69–7.71 m (1H). ^{13}C NMR spectrum, δ_{C} , ppm: 29.19, 36.58, 45.47, 65.33, 110.43, 112.45, 117.35, 121.02, 121.90, 126.20, 127.35, 127.60, 133.25, 140.23, 141.80, 142.85, 145.73. Found, %: C 72.8; H 6.9; N 12.1. $\text{C}_{21}\text{H}_{23}\text{N}_3\text{S}$. Calculated, %: C 72.8; H 6.8; N 12.0.

***N*-Benzyl-2-trimethylsilyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-amine (VII)** was obtained similarly. Yield 70%, mp 158°C. ^1H NMR spectrum, δ , ppm: 0.25 s (9H), 3.27–3.34 m (1H), 3.49–3.65 m (2H), 6.15 d (1H, $^3J_{\text{HH}}$ 4.36 Hz), 6.23–6.26 m (1H), 7.11–7.19 m (7H), 7.49–7.51 m (1H), 7.69–7.72 m (1H). ^{13}C NMR spectrum, δ_{C} , ppm: 2.23, 45.55, 65.82, 110.48, 117.35, 121.04, 121.99, 124.90, 126.22, 127.40, 127.60, 130.53,

133.28, 140.15, 142.28, 145.17. Found, %: C 65.6; H 6.4; N 11.4. $\text{C}_{20}\text{H}_{23}\text{N}_3\text{SSi}$. Calculated, %: C 65.7; H 6.3; N 11.5.

2-Phenyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-ol (VIII). 50 mL of distilled water preheated to 70°C was added to a solution of 0.0024 mol of compound I in 50 mL of ethanol under reflux. The resulting mixture was heated at 70°C during 2 min, slowly cooled to room temperature over 2 h and left at 4°C overnight. The obtained crystals were filtered off, washed with water, and dried. Yield 90%, mp 240°C. ^1H NMR spectrum, δ , ppm: 6.42 d (1H, $^3J_{\text{HH}}$ 4.4 Hz), 6.65 d. d (1H, $^3J_{\text{HH}}$ 8.7, $^3J_{\text{HH}}$ 4.4 Hz), 7.04 d (1H, $^3J_{\text{HH}}$ 8.7 Hz), 7.22–7.24 m (2H), 7.46–7.70 m (7H). ^{13}C NMR spectrum, δ_{C} , ppm: 73.47, 110.88, 116.60, 117.51, 121.45, 122.40, 126.08, 128.86, 129.40, 130.12, 133.30, 136.05, 142.74, 143.77. Found, %: C 68.5; H 4.4; N 9.9. $\text{C}_{16}\text{H}_{12}\text{N}_2\text{OS}$. Calculated, %: C 68.6; H 4.3; N 10.0.

2-tert-Butyl-4*H*-[1,3]thiazino[3,2-*a*]-benzimidazole-4-ol (IX) was obtained similarly. Yield 80%, mp 210°C. ^1H NMR spectrum, δ , ppm: 1.36 s (9H), 6.01 d (1H, $^3J_{\text{HH}}$ 4.4 Hz), 6.40–6.50 m (1H), 6.78 d (1H, $^3J_{\text{HH}}$ 7.3 Hz), 7.14–7.23 m (2H), 7.47–7.55 m (1H), 7.59–7.67 m (1H). ^{13}C NMR spectrum, δ_{C} , ppm: 29.27, 36.57, 73.12, 110.64, 112.74, 117.30, 121.10, 122.15, 133.25, 141.07, 142.80, 144.55. Found, %: C 64.5; H 6.1; N 10.8. $\text{C}_{14}\text{H}_{16}\text{N}_2\text{OS}$. Calculated, %: C 64.6; H 6.2; N 10.8.

ACKNOWLEDGMENTS

This work was financially supported by the Government of the Russian Federation (project no. 14.B25.310013).

REFERENCES

1. Stadnichuk, M.D., Khramchikhin, A.V., Pitserskaya, Yu.L., and Suvorova, I.V., *Russ. J. Gen. Chem.*, 1999, vol. 69, no. 4, p. 593.
2. Karpov, M.V., Khramchikhin, A.V., and Stadnichuk, M.D., *Russ. J. Gen. Chem.*, 2005, vol. 75, no. 3, p. 487. DOI: 10.1007/s11176-005-0257-3.
3. Shimizu, M., Hachiya, I., and Mizota, I., *Chem. Comm.*, 2009, p. 874. DOI: 10.1039/B814930E.
4. Karpov, M.V., Eremin, A.V., Fisher, A.I., Belyaev, A.N., and Stadnichuk, M.D., *Russ. J. Gen. Chem.*, 2011, vol. 81, no. 1, p. 128. DOI: 10.1134/S1070363211010208.
5. Racharlawar, S.S., Shankar, D., Karkhelikar, M.V., Sridhar, B., and Likhari, P.R., *J. Organomet. Chem.*, 2014, no. 757, p. 14. DOI: 10.1016/j.jorganchem-2014.01.028.

6. Karkhelikar, M.V., Racharlawar, S.S., Salián, S.M., Sridhar, B., and Likhar, P.R., *J. Organomet. Chem.*, 2012, no. 706–707, p. 128. DOI: 10.1016/j.jorganchem.-2012.02.009.
7. Trzhtsinskaya, B.V. and Abramova, N.D., *J. Sulfur Chem.*, 1991, vol. 10, no. 4, p. 389. DOI: 10.1080/01961779108048760.
8. Piterskaya, Yu.L., Suvorova, I.V., and Karpov, M.V., *Russ. J. Gen. Chem.*, 2012, vol. 82, no. 3, p. 515. DOI: 10.1134/S1070363212030280.
9. RU Patent 2061686, 1996.
10. RU Patent 2237663, 1999.
11. Karpov, M.V., Khramchikhin, A.V., and Stadnichuk, M.D., *Russ. J. Gen. Chem.*, 2005, vol. 75, no. 9, p. 1504. DOI: 10.1007/s11176-005-0457-x.
12. Karpov, M.V., Panina, N.S., Eremin, A.V., Stadnichuk, M.D., and Belyaev, A.N., *Russ. J. Gen. Chem.*, 2010, vol. 80, no. 6, p. 1193. DOI: 10.1134/S1070363210060241.
13. Usanova, E.A., Khramchikhin, A.V., and Stadnichuk, M.D., *Russ. J. Gen. Chem.*, 2000, vol. 70, no. 7, p. 1120.